

WWOA 2013

Advances In Control Strategies

October 2013



Outline

- Review of the Process Basics.
- Advances in Measurement Technology
- Advances in Control Technology
- Tighter Discharge Limits
- Cost Reduction (Chemical and Energy)
- Control Considerations
 - Control Fundamentals
 - Control Opportunities

Process Basics



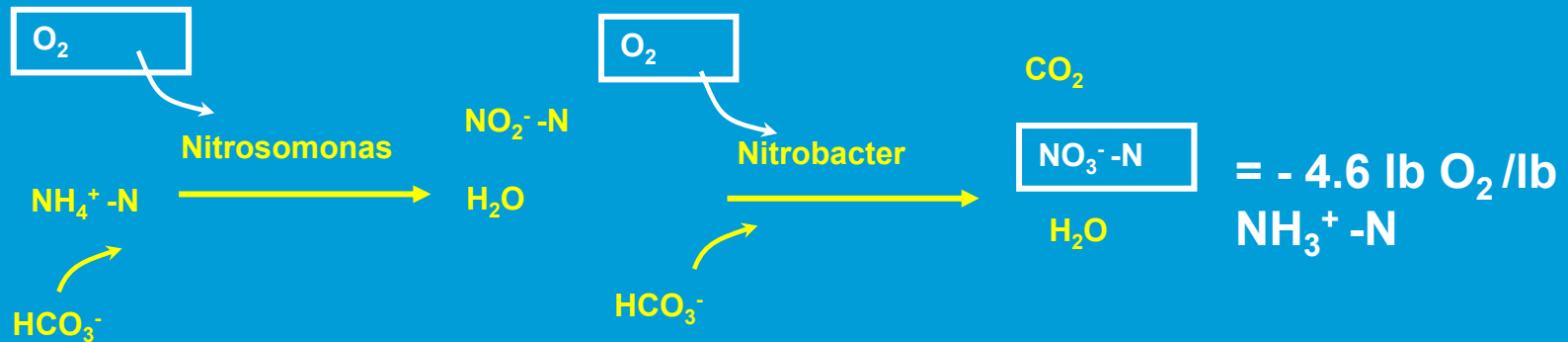


Process Fundamentals

- Special Focus On
 - Nitrification – Biological removal of ammonia
 - Denitrification – Biological removal of nitrate/nitrite
 - Biological Phosphorus Removal
 - Chemical Phosphorus Removal
- Other Processes
 - BOD Removal
 - Disinfection

Nitrification and Denitrification

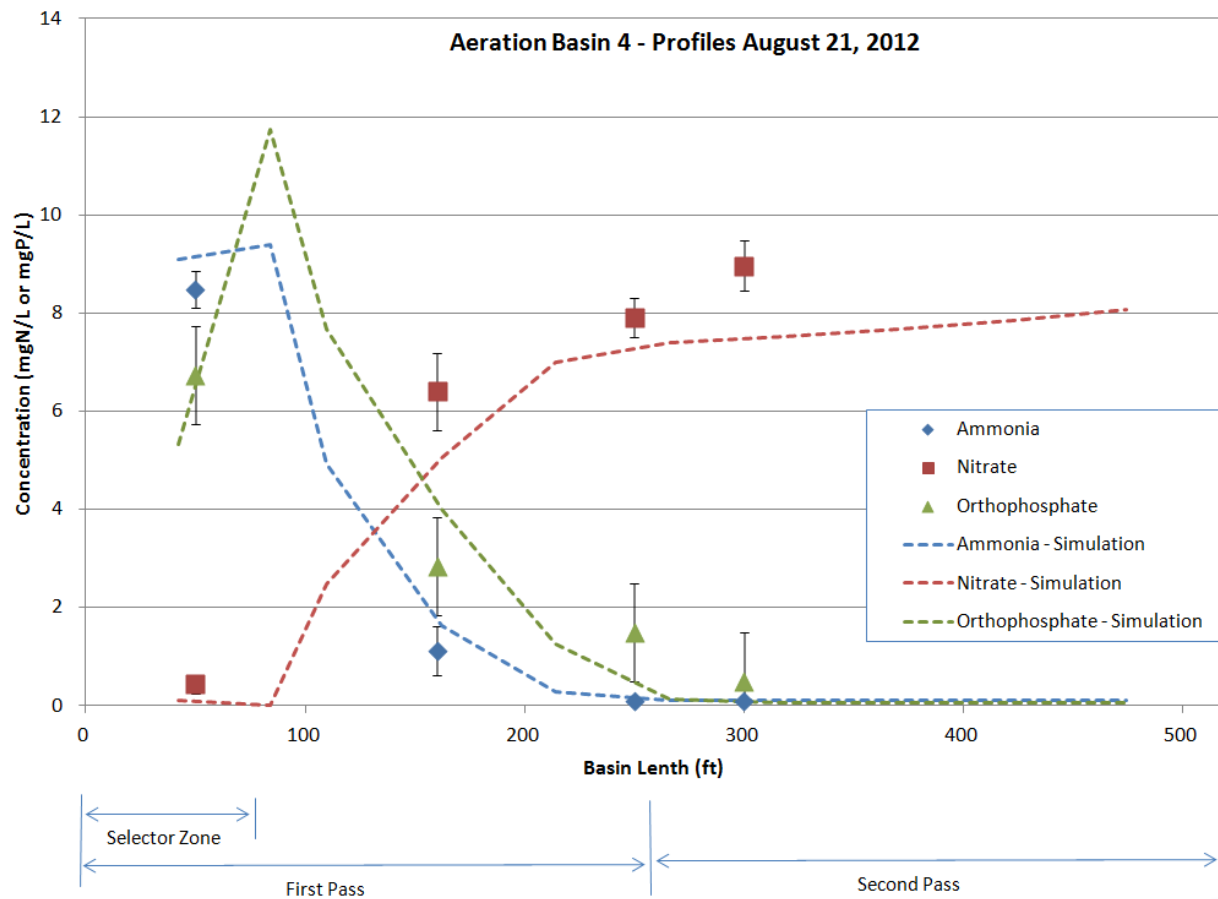
Nitrification



Denitrification



Plug Flow Nitrification Trend



Advances in Measurement Technology

- Online Analyzers
 - Lower Capital Cost
 - More Reliable with Less Maintenance
 - Lower Consumable Costs
 - Better Accuracy

Parameters

*Limited Range and Application

Available Parameters

Parameters ASA Pub #	Parameter	C h e m S c a n	MODERNWATER
41	Ammonia, Free	X	
105	Ammonia, Total	X	
158	Antimony		X
	Aquatic Humic Substances (AHS)	X	
	Arsenic		X
136	Barium	X	
	Cadmium		X
102	Chloramine	X	
40	Chlorine, Free	X	
121	Chlorine, Total	X	
133	Chrome VI	X	X
	Cobalt		X
30	COD	X	
52	Color, Aparent	X	
125	Color, True	X	
06	Copper	X	X
150	Dissolved Organic Carbon (DOC)	X	
	Gold		X
85	Hardness, Total	X	
100	Humic Matter	X	
106	Iodine	X	
141	Iron, Ferric	X	
43	Iron, Ferrous	X	
120	Iron, Total	X	X
	Lead		X
134	Manganese	X	X
	Mercury, Dissolved		X
123	Molybdate	X	X
122	Monochloramine	X	
08	Natural Organic Matter (NOM)	X	
	Nickel	X	X
06	Nitrate	X	
57	Nitrite	X	
130	Nitrogen, Total	X	
124	Nitrogen, Total Oxidized	X	
	Nutrients, Multiple	X	
	Organic Acids	X	
157	Ozone, Dissolved	X	
	Palladium		X
42	Percent Transmittance	X	
40	Phosphate, ortho	X	
124	Phosphate, poly	X	
131	Phosphorous, Total	X	
149	Polymers, Synthetic	X	
09	Potassium Permanganate	X	
	Selenium		X
144	Silica	X	
	Silver		X
107	Solids Correlation	X	
127	Specific Absorbance	X	
128	Spectrum Matching	X	
125	Sulfate	X	
168	Sulfite, Dechlorination	X	
101	Tannin	X	
	Tellurium		X
	Thallium		X
	Tin		X
126	TOC	X	
	Toxicity, Specific Metals		X
120	Triazole	X	
07	Turbidity*	X	
	Zinc		X

*Limited Range and Application

Highlighted In-Stream Technologies

- Ammonia
- Nitrate
- Ortho-Phosphate
- Oxidation Reduction Potential (ORP)





History of Control Technology

- Programmable Logic Controllers (PLC)
 - Developed for GM in the 1970s.
 - Meant as Relay Replacement Units (Digital)
 - Originally Analog was expensive and cumbersome.
- HMI Software
 - Follow the Development of PLC by about 10 years.
 - Quickly caught up with current technology.
- Distributed Control Systems (DCS) was cost prohibitive except in the largest facilities.



Advances in Control Technology

- PLC Hardware
 - Analog Modules are less expensive.
 - Analog Modules have enough resolution to handle Process Control Applications
 - Processor have the memory and speed to handle Process Control.
- PLC Software
 - Data Structure to Support Process Control Application.
 - Additional Languages to support Process Control.
- PLC Manufactures are determined to serve this market

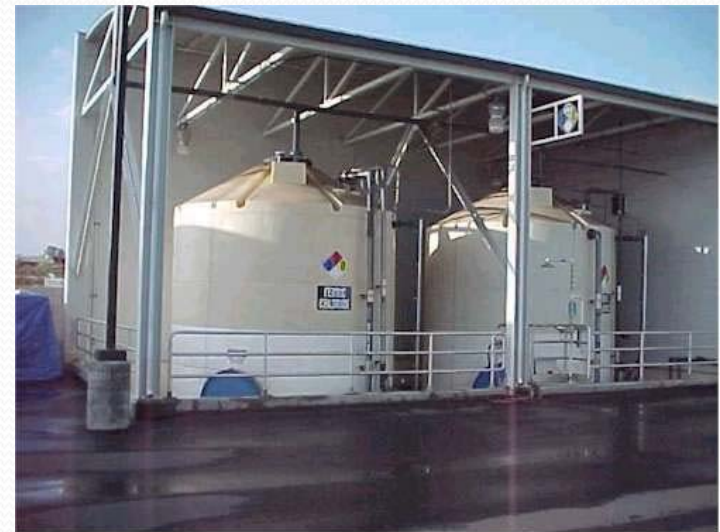


Tighter Discharge Limits

- EPA and DNR Continue to Dictate Tighter Limits on Phosphorus.
- “The DNR estimates up to 163 municipal plants may need new filtration systems that could run a total of \$300 million to \$1.13 billion.”
- Very little point in belaboring this issue.

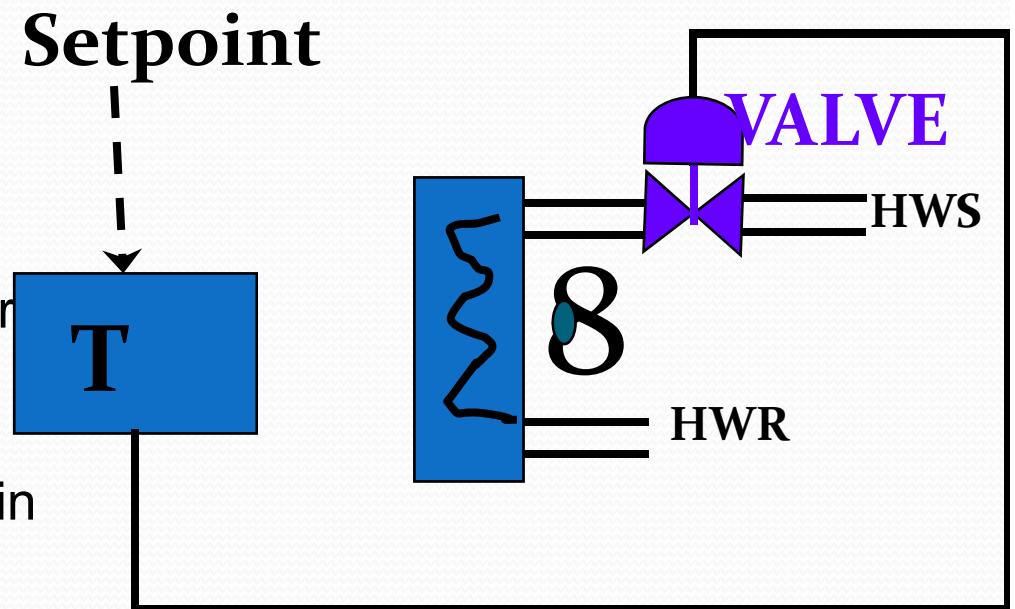
Cost Reduction

- It is oblivious if there was a magic money tree it is gone.
- Energy savings has always made sense when cost effective.
- Lowering Chemical Costs makes sense.
- Lowering Chemical Usage minimizes the chemical byproduct effects on waterways



Control Fundamentals

- To Control a Parameter it Must be Measured.
- To Control a Parameter there must be a Controllable Device. (VFD, Valve, Gate Position, etc)
- More Information is always better than less.
- Math and Algorithm Capabilities in PLCs are unlimited. (Add, Sub, Mult, Divide, Square Root, All Trig Function, Log, Ln and Compute)



Algorithm Example

CALCULATED WIND
CHILL TEMPERATURE

CPT

Compute
Dest

AD_WC_TEMPERATURE_OUTPUT
51.139107 ←

Expression $35.74 + 0.6215 * \text{AD_AIR_TEMPERATURE.O.AS_OUTPUT} - 35.57 * \text{AD_WIND_SPEED.O.AS_OUTPUT}^{0.16} + 0.4275 * \text{AD_AIR_TEMPERATURE.O.AS_OUTPUT} * \text{AD_WIND_SPEED.O.AS_OUTPUT}^{0.16}$

Control Opportunities

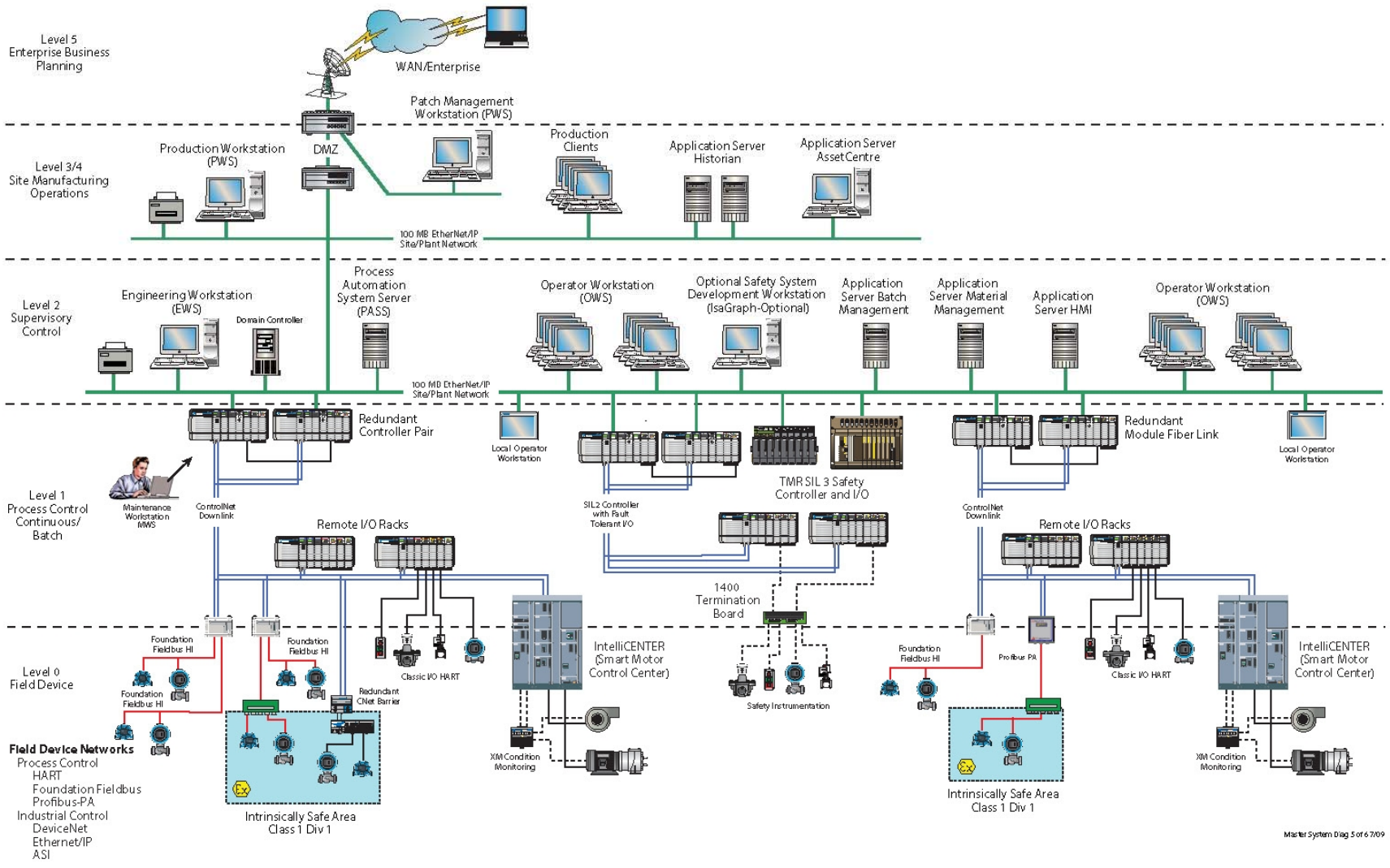


Overview of Process Control Systems

- Field Instrumentation
- Control Devices (usually PLCs)
- Human Machine Interface (Hardware and Software)
- Process Historians
- Reporting Systems

Rockwell Automation
Process Control Multi Server/Multi Client System

PlantPax
 Process Automation System



Data, Information, Knowledge and *Control*

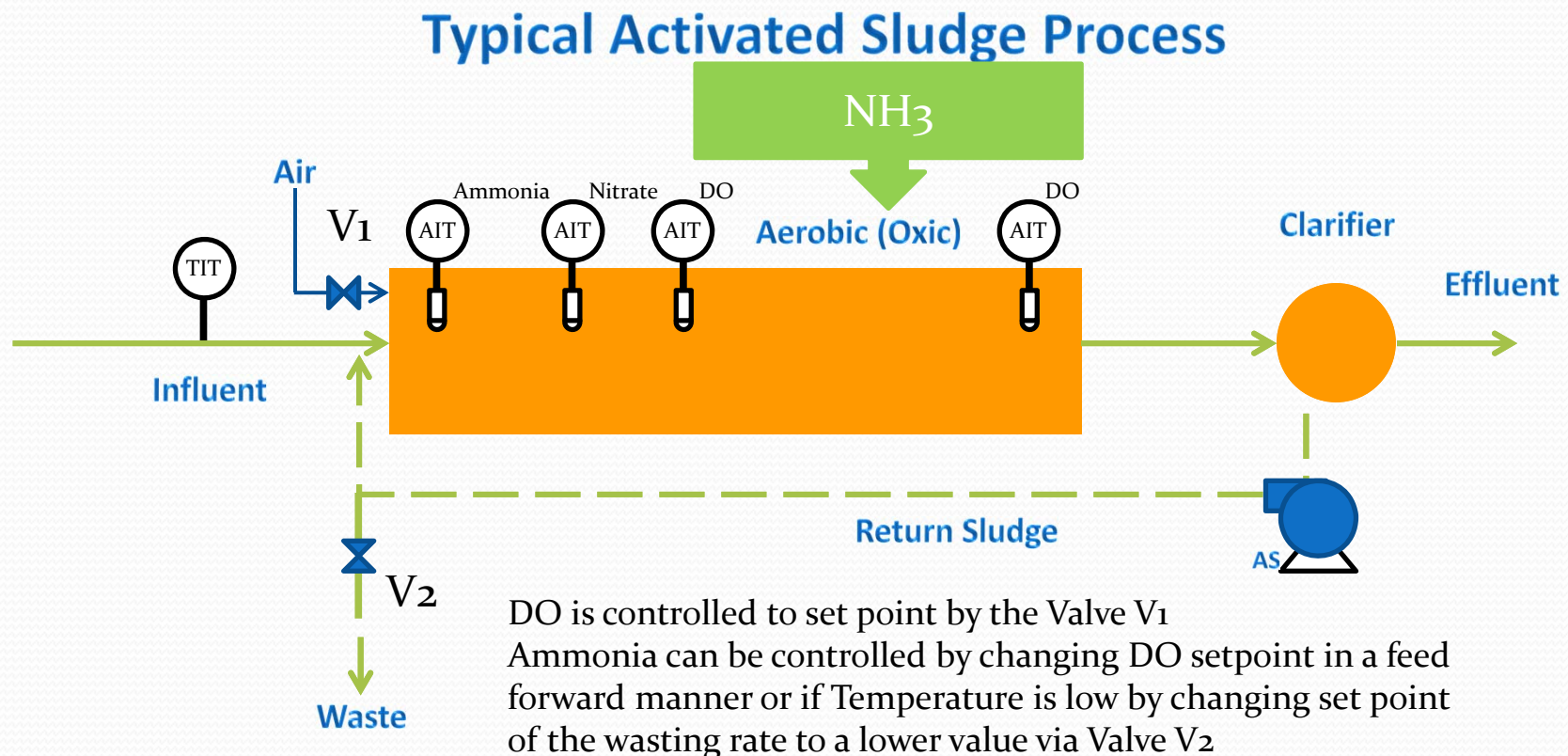
- Data – Your daily log sheets
- Information – Charting or Trending the log sheets
- Knowledge - Seeing patterns that answers the How questions
- Control - Implementing Systems to Control what we now have Knowledge of.

Key

- Nitrification – Ammonia Removal
- Denitrification – Nitrate Removal
- BPR – Biological P Removal
- CPR – Chemical P Removal

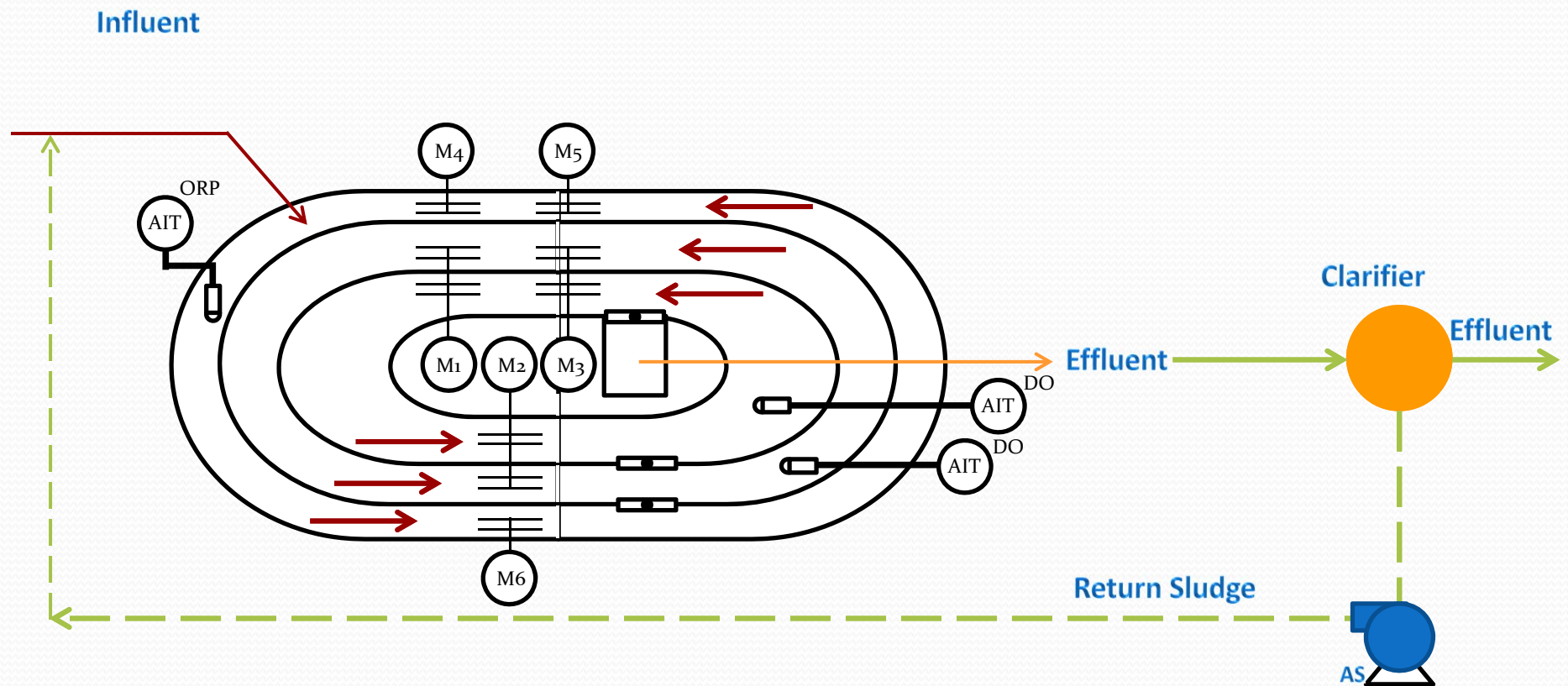


Plug Flow Process with Nitrification



Simple BPR Process

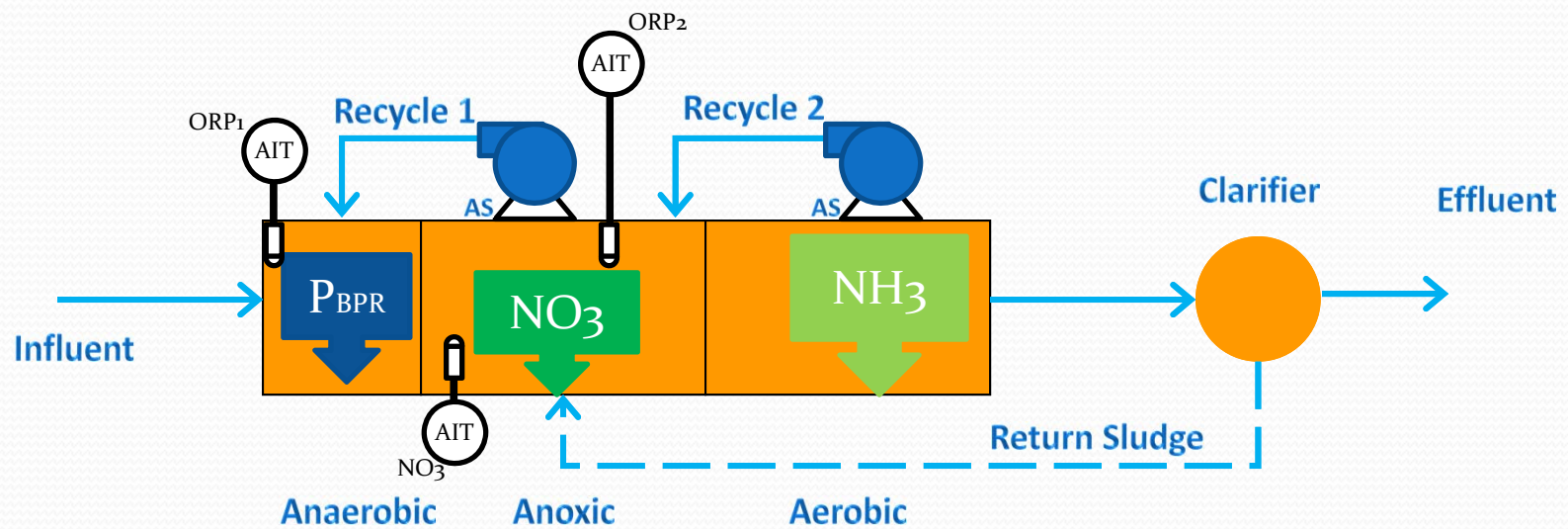
Simple BPR Process (A/O Process)



DO controlled by changing the speed of Aerator Motor M_{1,2,3}
ORP controlled by changing the speed of Aerator M_{4,5,6}
ORP could be affected by the return rate.

BPR Process

UCT Process



ORP₁ can be controlled by changing the speed of Recycle Pump 1
ORP₂ can be controlled by changing the speed of Recycle Pump 2
Nitrate can be controlled by changing Recycle Pump 2 Rate with
the nitrate being the master variable in this zone.

The Phosphorus Calculation

- By measuring the mg/l of Phosphorus and the Total Influent Flow to the Plant, the Total Phosphorus entering the plant can be determined:

$$\text{Total P (lb/day)} = \text{Phosphorus mg/l} * \text{Plant Flow MGD} * 8.34$$

- Calculate the pounds of Fe required by the equation (based on 1:1 molar ratio):

$$\text{Fe Req'd (lb/day)} = \text{Total P (lb/day)} * \frac{1 \text{ mole P}}{31 \text{ lbs P}} * \frac{55.85 \text{ lb Fe}}{1 \text{ mole Fe}}$$

The Final Calculation

- Include Nullification Ratio of Fe to Phosphorus

1.6 Moles Fe/1 Mole Phosphorus (Typical range 1.6-2 in an activated sludge plant)

$$\text{Fe Req'd (adjusted) (lb/day)} = 1.6 * \text{Fe Req'd (lb/day)}$$

- Final Calculation to Determine FeCl_3 Flow Rate:

$$\begin{aligned} \text{FECL}_3 \text{ Flow Rate (gpd)} &= \text{Fe Req'd (adjusted) (lb/day)} \\ &\quad * 162 \text{ lb FeCl}_3 / 55.85 \text{ lb Fe} * 1 \text{ lb solution} / 35\% \text{ solution} \\ &\quad * 1 \text{ gallon} / (8.34 * \text{SG}) \end{aligned}$$

$$\text{FeCl}_3 \text{ Solution per hour} = \text{Flow Rate (gpd)} / 24$$



The Cost Savings

- Flow Pace Only vs. Phosphorus Calculation
- Plant under normal conditions 12 MGD and 7 mg/l P
- Flow rate in a Rain Event goes to 30 MGD at 2.8 mg/l. Gallons of Solution goes to 158 gallons per hour but should be only 67 gallons per hour. Cost per hour lost is \$113.
- Night time flows go to 6 MGD and Phosphorus goes to 5 mg/l. Flow rate of Solution goes to 31 gallons per hour but should go to 23 gallons per hour. Cost per hour lost is \$10.



Questions?